
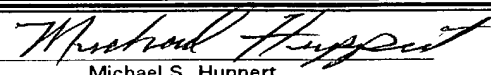


FORM PTO 1390 (REV. 5-93) US DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE		ATTORNEY DOCKET NUMBER 2002_0266A
TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. §371		U.S. APPLICATION NO. 107-070374
International Application No. PCT/NO00/00294	International Filing Date September 8, 2000	Priority Date Claimed September 10, 1999
Title of Invention A CARBON ELECTRODE AND A METHOD FOR PRODUCING SUCH AN ELECTRODE		
Applicant(s) For DO/EO/US Egil LUNDBERG		
Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:		
1. <input checked="" type="checkbox"/> This is a <b>FIRST</b> submission of items concerning a filing under 35 U.S.C. §371. 2. <input type="checkbox"/> This is a <b>SECOND</b> or <b>SUBSEQUENT</b> submission of items concerning a filing under 35 U.S.C. §371. 3. <input checked="" type="checkbox"/> This express request to begin national examination procedures (35 U.S.C. §371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. §371(b) and PCT Articles 22 and 39(1). 4. <input checked="" type="checkbox"/> A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date. 5. <input checked="" type="checkbox"/> A copy of the International Application as filed (35 U.S.C. §371(c)(2)) a. <input checked="" type="checkbox"/> is transmitted herewith (required only if not transmitted by the International Bureau). <b>ATTACHMENT A</b> b. <input type="checkbox"/> has been transmitted by the International Bureau. c. <input type="checkbox"/> is not required, as the application was filed in the United States Receiving Office (RO/US) 6. <input type="checkbox"/> A translation of the International Application into English (35 U.S.C. §371(c)(2)). 7. <input type="checkbox"/> Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. §371(c)(3)). a. <input type="checkbox"/> are transmitted herewith (required only if not transmitted by the International Bureau). b. <input type="checkbox"/> have been transmitted by the International Bureau. c. <input type="checkbox"/> have not been made; however, the time limit for making such amendments has NOT expired. d. <input type="checkbox"/> have not been made and will not be made. 8. <input type="checkbox"/> A translation of the amendments to the claims under PCT Article 19. 9. <input type="checkbox"/> An oath or declaration of the inventor(s) (35 U.S.C. §371(c)(4)). 10. <input type="checkbox"/> A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. §371(c)(5)). Items 11. to 14. below concern other document(s) or information included: 11. <input checked="" type="checkbox"/> An Information Disclosure Statement under 37 CFR 1.97 and 1.98. - <b>ATTACHMENT C</b> 12. <input type="checkbox"/> An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included. 13. <input checked="" type="checkbox"/> A <b>FIRST</b> preliminary amendment. <b>ATTACHMENT D</b> <input type="checkbox"/> A <b>SECOND</b> or <b>SUBSEQUENT</b> preliminary amendment. 14. <input checked="" type="checkbox"/> Other items or information: Unexecuted Declaration and Power of Attorney along with cover letter - <b>ATTACHMENT B</b> Form PCT/IB/304 - <b>ATTACHMENT E</b> International Preliminary Examination Report - <b>ATTACHMENT F</b>		

U.S. APPLICATION NO. <b>10/070374</b> NEW		INTERNATIONAL APPLICATION NO. PCT/NO00/00294		ATTORNEY'S DOCKET NO. 2002 0266A					
15. <input checked="" type="checkbox"/> The following fees are submitted  <b>BASIC NATIONAL FEE (37 CFR 1.492(a)(1)-(5)):</b> Neither international preliminary examination fee nor international search fee paid to USPTO and International Search Report not prepared by the EPO or JPO ..... \$1040.00 International Search Report has been prepared by the EPO or JPO ..... \$ 890.00 International preliminary examination fee not paid to USPTO but international search paid to USPTO ..... \$ 740.00 International preliminary examination fee paid to USPTO but claims did not satisfy provisions of PCT Article 33(1)-(4) ..... \$ 690.00 International preliminary examination fee paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) ..... \$ 100.00  <b>ENTER APPROPRIATE BASIC FEE AMOUNT =</b>				<table border="1" style="width:100%; border-collapse: collapse;"> <tr> <th style="width:50%;">CALCULATIONS</th> <th style="width:50%;">PTO USE ONLY</th> </tr> <tr> <td style="height: 100px;"></td> <td></td> </tr> </table>		CALCULATIONS	PTO USE ONLY		
CALCULATIONS	PTO USE ONLY								
Surcharge of \$130.00 for furnishing the oath or declaration later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492(e)).				\$					
Claims	Number Filed	Number Extra	Rate						
Total Claims	13 -20 =	-0-	X \$18.00	\$					
Independent Claims	2 - 3 =	-0-	X \$84.00	\$					
Multiple dependent claim(s) (if applicable)			+ \$280.00	\$					
<b>TOTAL OF ABOVE CALCULATIONS =</b>				\$1,040.00					
<input type="checkbox"/> Small Entity Status is hereby asserted. Above fees are reduced by 1/2.				\$					
<b>SUBTOTAL =</b>				\$1,040.00					
Processing fee of \$130.00 for furnishing the English translation later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492(f)).				+					
<b>TOTAL NATIONAL FEE =</b>				\$1,040.00					
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40 per property +				\$					
<b>TOTAL FEES ENCLOSED =</b>				\$1,040.00					
				Amount to be refunded	\$				
				Amount to be charged	\$				
a. <input checked="" type="checkbox"/> A check in the amount of \$1,040.00 to cover the above fees is enclosed. A duplicate copy of this form is enclosed. b. <input type="checkbox"/> Please charge my Deposit Account No. 23-0975 in the amount of \$_____ to cover the above fees. A duplicate copy of this sheet is enclosed. c. <input type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 23-0975.									
<b>NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b))          must be filed and granted to restore the application to pending status.</b>									
19. CORRESPONDENCE ADDRESS   <div style="text-align: center;">   <b>000513</b>          PATENT TRADEMARK OFFICE       </div>			By:  Michael S. Huppert, Registration No. 40,268  WENDEROTH, LIND & PONACK, L.L.P. 2033 "K" Street, N.W., Suite 800 Washington, D.C. 20006-1021 Phone: (202) 721-8200 Fax: (202) 721-8250  March 6, 2002						

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of :  
Egil LUNDBERG : Attn: BOX PCT  
Serial No. NEW : Docket No. 2002\_0266A  
Filed March 5, 2002 :

A CARBON ELECTRODE AND A METHOD  
FOR PRODUCING SUCH AN ELECTRODE  
[Corresponding to PCT/NO00/00294  
Filed September 8, 2000]

THE COMMISSIONER IS AUTHORIZED  
TO CHARGE ANY DEFICIENCY IN THE  
FEES FOR THIS PAPER TO DEPOSIT  
ACCOUNT NO. 23-0975

PRELIMINARY AMENDMENT

Assistant Commissioner for Patents,  
Washington, DC 20231

Sir:

Prior to initial examination of the above-identified application, kindly amend  
the application as follows:

IN THE CLAIMS:

Kindly cancel original claims 1-5 without prejudice or disclaimer recited  
thereof.

Kindly add the following new claims:

6.(NEW) A method for producing a carbon electrode in which a "green" mass  
comprising particle material containing carbon and a binder undergoes a moulding process  
which causes the mass to be exposed to externally forced compression in one or more  
directions and to be subjected to a calcination process before use, characterised in that

the carbon electrode is arranged so that, when it is in use, the dominant direction of electric current will mainly be oriented so that it does not coincide with the direction(s) of the forced compression.

7.(NEW) A method **in accordance with claim 6** for production of a carbon electrode, more precisely an anode for use in an electrolysis cell of Hall-Hérault type in which the anode is made with at least one recess for fixing to an anode suspender, characterised in that each recess is arranged directionally so that it mainly coincides with a direction mainly perpendicular to the direction(s) of the forced compression.

8.(NEW) A method **in accordance with claim 7**, characterised in that the carbon electrode is calcinated before the recesses are arranged.

9.(NEW) A method **in accordance with claim 8**, characterised in that the recesses are arranged by a mechanical milling or drilling process.

10.(NEW) A carbon electrode produced from a "green" mass comprising particle material containing carbon and a binder where the green mass is exposed to externally forced compression in one or more directions and the carbon electrode is subjected to a calcination process before use, characterised in that at least one electrical connector is arranged in the electrode in such a manner that the dominant direction of electric current

in relation to the carbon electrode, when it is in use, mainly does not coincide with the direction(s) of the forced compression.

11.(NEW) A carbon electrode **in accordance with claim 10**, more precisely an anode for use in an electrolysis cell of Hall-Héroult type in which the anode is made with at least one recess for fixing to an anode suspender, characterised in that each recess is arranged in such a manner with respect to the extension of its depth into the anode so that this direction mainly coincides with a direction substantially perpendicular to the direction(s) of the forced compression.

12.(NEW) A carbon electrode **in accordance with claim 11**, characterised in that it is calcinated before the recesses are arranged.


13.(NEW) A carbon electrode **in accordance with claim 12**, characterised in that the recesses are arranged by drilling or by milling the calcinated carbon material.

**REMARKS**

The present Preliminary Amendment is submitted to cancel original claims 1-5 and add new claims 6-13. Note that the new claims are presented in order to incorporate the amendments filed in the international application.

Respectfully submitted,

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March 6, 2002

3/pvt

**A Carbon Electrode and a Method for Producing such an Electrode**

The present invention concerns an improved carbon electrode and a method for producing a carbon electrode. Carbon electrodes, particularly anodes, produced in accordance with the present invention may expediently be used in connection with electrolytic production of aluminium in accordance with the Hall-Héroult process involving pre-baked anodes.

The present invention is based on the observed fact that several physical properties of carbon electrodes will be directional on the basis of the moulding process used. This applies, among other things, to electrodes moulded by vibration moulding, for which differences can be demonstrated between the vertical and horizontal directions.

A common method for producing anodes for use for aluminium production is vibration moulding of a "green" mass (a viscous, ductile mass containing carbon particles and binder) in a mould consisting of a box open at the top which has a plumb or a heavy lid designed to slide downwards along the inner walls of the box. Nipple holes or recesses in the anode for fixing it to an anode suspender are usually created by the plumb having downward-facing projections which extend down into the mass. The creation of anodes in this way means that the orientation of the recesses corresponds to the vibration direction (vertical direction). One disadvantage of the above production method is that the physical properties of the anode cannot be exploited in an optimized manner because of limitations in the actual production method.

One explanation of the directional difference may be related to how particles inside the material move during the moulding operation. For example, the external geometric dimensions of the mass during vibration will be reduced in the vertical direction, while the dimensions will remain virtually constant in the horizontal direction. Another reason may be that the mass which is vibrated contains carbon particles which, to a large extent, have the form of oblong flakes. During the vibration of the "green" mass, the flakes will tend to be adjusted so that their centre of gravity is located on the lowest possible vertical level. This means that there may be more interfaces between the carbon particles in the vertical direction than in the horizontal direction, which is assumed to be a dominant factor regarding the fact that the physical properties such as mechanical strength, electrical resistance, thermal properties, etc. are directional in relation to the moulding process used.

With the present invention, it has become possible for a carbon electrode to be produced so that its physical properties can be utilised optimally. With the present invention, a carbon electrode will be produced with reduced electrical resistance and more favourable thermal conductivity properties. With the present invention, it will also be possible to use  
5 simpler materials than previously without having to reduce the requirements for the properties stated.

The present invention will be described in the following using examples and figures, where:

10

Figure 1 shows the physical properties of a carbon electrode.

Figure 2 shows how sampling is done in relation to a carbon electrode.

Figure 3 gives a graphic presentation of the difference between vertical and horizontal resistance in a carbon electrode.

15 Figure 4 shows a comparison between density and resistance in a carbon electrode.

The vibration direction will be called the vertical direction (V) in the following. Correspondingly, the horizontal direction (H) is perpendicular to this.

20 Two core samples were drilled out in both directions from 9 areas in typical carbon electrodes, see Figure 2. The areas were in a plane 200 mm above the underside of the carbon electrode, i.e. where the wear surface is located after half the operating life period in an electrolysis process. The points of intersection between this and three vertical planes longitudinally to and three vertical planes transversely to the carbon describe  
25 where the samples were taken. The vertical samples had their centre axis in the intersection between the longitudinal and transverse planes and in such a way that the horizontal plane intersected them at half their height. The horizontal samples had their centre axis in the horizontal plane and as close to the others as possible.

30 The samples were tested in relation to a number of parameters, which are shown in Figure 1:

- Reactivity in carbon dioxide,  $R_{CO_2}$

Expresses the carbon electrode's (anode's) tendency to react with carbon dioxide at  
35  $960^{\circ}C$ . A high value of this means high reactivity.

- Soot index,  $S_{CO_2}$



Expression of selective reaction with carbon dioxide which results in loose particles (soot) in the electrolysis bath.

- Density (unit weight, volume weight)

5 Calculated on the basis of the sample's weight and external dimensions.

- Specific electrical resistance

Calculated on the basis of the measured voltage drop over the sample and its cross-section and length.

10

- Young's modulus, YM

Modulus of elasticity, determined by measuring compression in a compression strength test.

15 - Compression strength, CS

Calculated on the basis of the force applied in connection with compression to break.

- Air permeability, Perm

Expression of continuous pores. A high value corresponds to open material.

20

- Coefficient of thermal expansion, CTE

Linear expansion as a result of change in temperature.

- Reactivity in air,  $R_{AIR}$

25 Expresses the carbon electrode's (anode's) tendency to react with air at 525°C. A high value corresponds to high reactivity.

- Porosity, Por

Total porosity based on image analysis.

30

The table in Figure 1 indicates typical values for the horizontal and vertical directions.

The permeability is slightly higher in the horizontal direction than in the vibration direction.

This corresponds with the porosity determined in samples from the centre axis. However,

35 it has not been demonstrated that this can produce a noticeable increase in the internal CO<sub>2</sub> reactivity in the carbon.

The other direction-dependent parameters, resistance (converted into thermal conductivity), YM, CS and CTE are subject to considerations of thermal stress. Modelling tests with the values in question give no reason to expect significant changes in these forces in the carbon electrode (anode).

5

Figure 3 shows the directional difference between vertical and horizontal specific electrical resistance in each of the 9 sample points, expressed in a bar chart.

It can usually be observed that density and resistance will correspond well (high density produces low resistance), in particular when the raw material and process are generally the same and with standard sampling, i.e. in the vibration direction. The table in Figure 4 shows this, but also that this is not so marked when the resistance is measured in the H direction. The latter tendency probably increases as the density decreases.

15 The last line in the table in Figure 4 indicates that the correlation between density and the difference in resistance between the directions is low, at least for the anode quality in question.

If the manufacturing process is such that the nipple holes in an anode are created entirely after moulding, for example by milling or by drilling nipple holes after calcination, it is possible to choose the side on which they are to be placed. It is thus possible to benefit from the anisotropy by ensuring that the direction of electric current flow in the electrolysis coincides with the H direction in connection with vibration. In accordance with commonly used vibration/compression techniques, this will imply that the nipple holes are arranged substantially perpendicular to the direction of vibration/compression of the electrode in its "green state".

It should be understood that electrodes produced in a way where the "green mass" is compressed merely in a static manner or by extruding techniques may in the same manner as described above have directional properties which can be exploited in accordance with the present invention.

The size of the power saving which can be achieved with this will depend on how the anode is produced. On the basis of a typical anode as described earlier, the total energy saving will be 0.31% on the basis of the below conditions:

Total voltage drop over cell

: 4 V

Average voltage drop over anode	: 150 mV
Difference in specific electrical resistance	: 4.5 $\mu\Omega\text{m}$
Power consumption	: 14 kWh/kg Al
Reduction in resistance in the carbon itself	: 8.3 %

5

The present invention thus offers a considerable potential for savings in the form of reduced power consumption. The present invention will also make it possible for the carbon electrode, in connection with vibration, to be given a more precise height as the nipples in the finished anode are innstalled in a direction in which the geometric  
10 dimensions of the mass during tamping/vibration are kept constant.

### Claims

- 5 1. A method for producing a carbon electrode in which a "green" mass comprising particle material containing carbon and a binder undergoes a moulding process which causes the mass to be exposed to externally forced compression in one or more directions and to be subjected to a calcination process before use,  
characterised in that  
10 the carbon electrode is arranged so that, when it is in use, the dominant direction of electric current will mainly be oriented so that it does not coincide with the direction(s) of the forced compression.
2. A method in accordance with claim 1 for production of a carbon electrode, more  
15 precisely an anode for use in an electrolysis cell of Hall-Héroult type in which the anode is made with at least one recess for fixing to an anode suspender,  
characterised in that  
each recess is arranged directionally so that it mainly coincides with a direction  
mainly perpendicular to the direction(s) of the forced compression.
- 20 3. A method in accordance with claim 2,  
characterised in that  
the carbon electrode is calcinated before the recesses are arranged.
- 25 4. A method in accordance with claim 3,  
characterised in that  
the recesses are arranged by a mechanical milling or drilling process.
5. A carbon electrode produced from a "green" mass comprising particle material  
30 containing carbon and a binder where the green mass is exposed to externally forced compression in one or more directions and the carbon electrode is subjected to a calcination process before use,  
characterised in that  
the dominant direction of electric current in relation to the carbon electrode,  
35 when it is in use, mainly does not coincide with the direction(s) of the forced compression.

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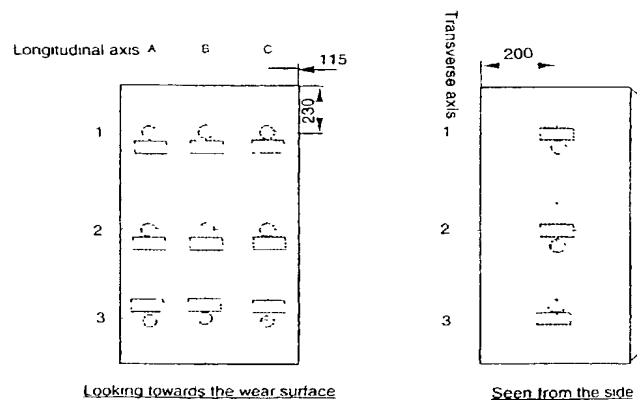
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**Published:**

— With international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette

(54) Title: A CARBON ELECTRODE AND A METHOD FOR PRODUCING SUCH AN ELECTRODE



(57) Abstract: The present invention concerns an improved carbon electrode and a method for producing such a carbon electrode. In particular, the present invention relates to anodes for use in connection with electrolytic production of aluminium in accordance with the Hall-Héroult process. The anisotropy in a vibrated carbon anode results in partially significant differences in the physical properties depending on how the samples are oriented in relation to the vibration direction, in particular with regard to electrical resistance. For a tested, typical quality electrode, the resistance perpendicular to the vibration direction is 8.3 % lower than in the vibration direction. If this is utilised by placing the nipple or suspension hanger holes so that the direction of electric current flow when the electrode is in use in the electrolysis is substantially 90° to the vibration/compression direction, this can produce a reduction of approximately 0.31 % in power consumption.

Direction	$R_{CO_2}$ mg/cm <sup>2</sup> /h	$S_{CO_2}$ %	Density mg/cm <sup>3</sup>	Resistance $\mu\Omega m$	YM MPa	Com-pression strength MPa	Perm nPm	Thermal expansion 1/K·10 <sup>6</sup>	RAIR mg/cm <sup>2</sup> /h	Por %
H	24.0	3.5	1.570	49.8	10219	42.0	0.8	4.1	25.4	21.6
V	23.8	3.5	1.571	54.3	8867	39.3	0.5	4.6	27.5	19.6

Fig. 1

2/3

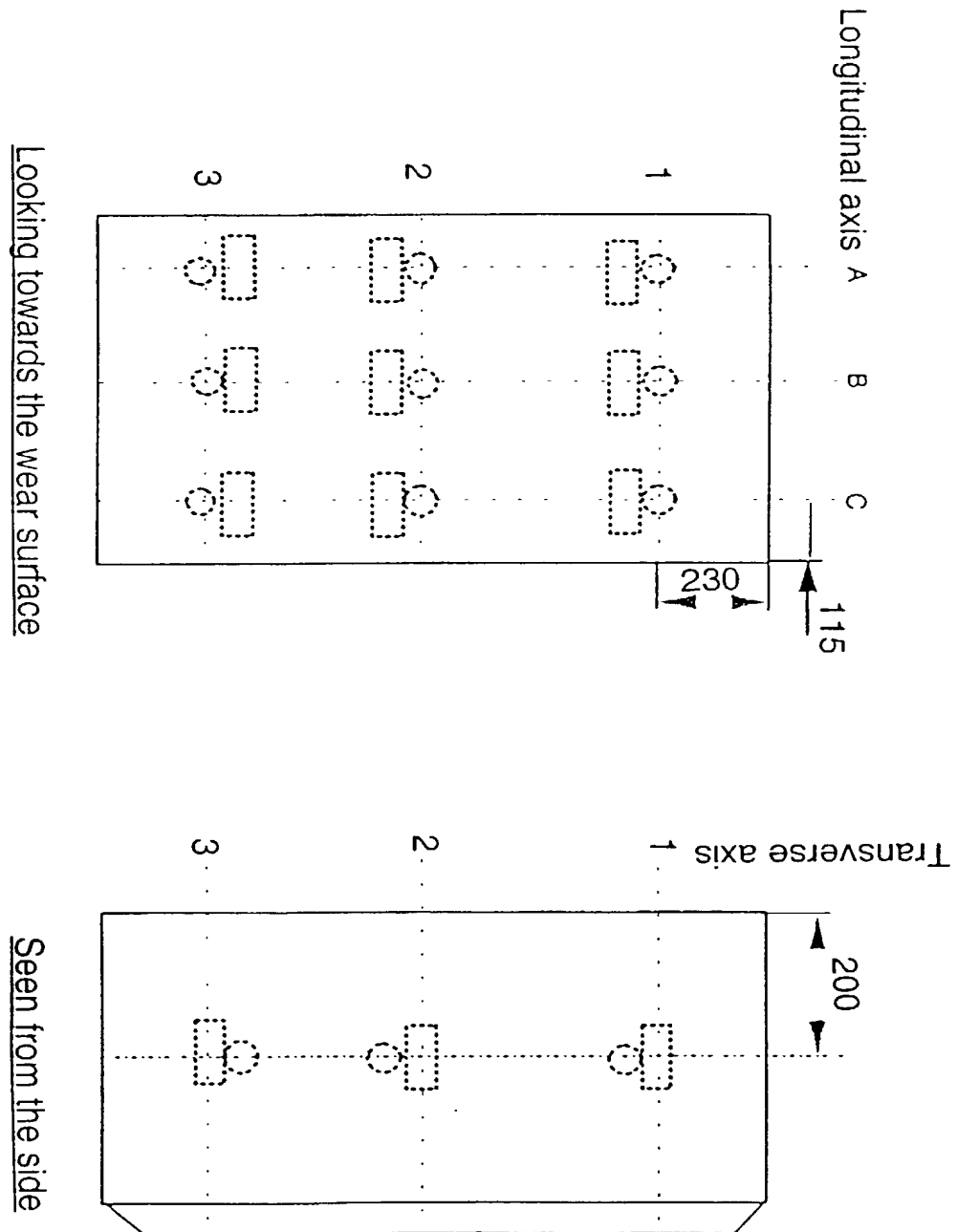


Fig. 2

3/3

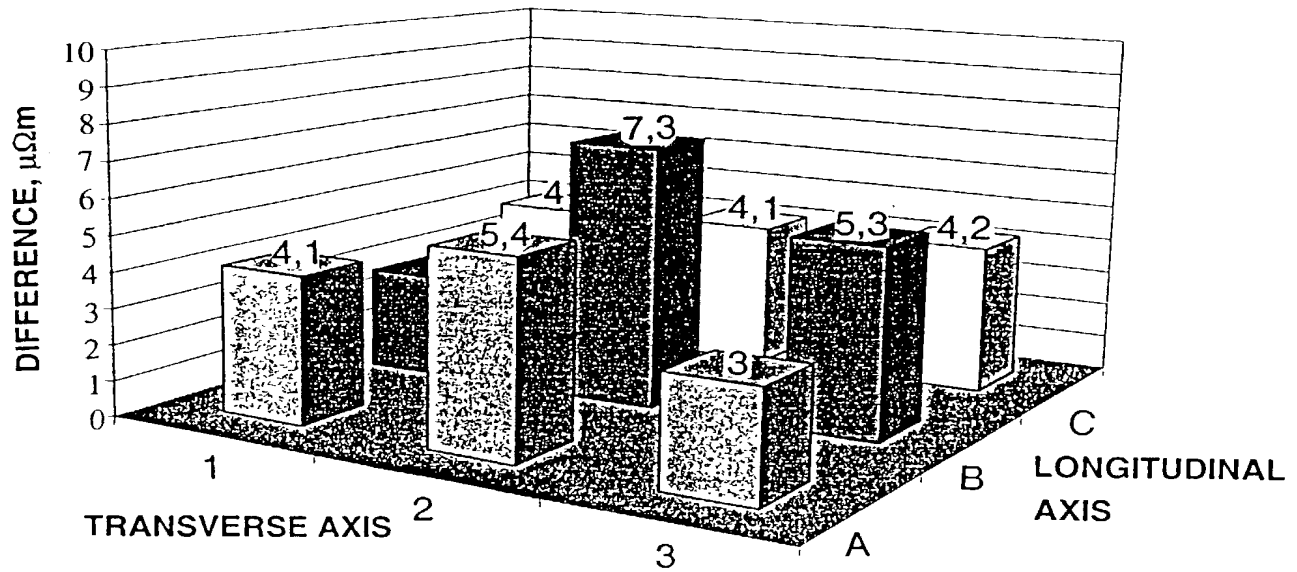


Fig. 3

PARAMETERS	COEFFICIENTS OF CORRELATION
Density - resistance H	-0.78
Density - resistance V	-0.86
Density - resistance H-V	-0.35

Fig. 4



Rev. 5/30/01

Effective March 1998

**DECLARATION AND POWER OF ATTORNEY FOR U.S. PATENT APPLICATION**

( ) Original ( ) Supplemental ( ) Substitute (x) PCT ( ) Design

As a below named inventor, I hereby declare that my residence, post office address and citizenship are as stated below next to my name; that I verily believe that I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural inventors are named below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

Title: "A carbon electrode and a method for producing such an electrode"

(derived from PCT/NO00/00294)

of which is described and claimed in:

- ( ) the attached specification, or  
 ( ) the specification in the application Serial No. \_\_\_\_\_ filed \_\_\_\_\_;  
     and with amendments through \_\_\_\_\_ (if applicable), or  
 (x) the specification in International Application No. PCT/NO00/00294, filed September 8, 2000, and as amended  
     on \_\_\_\_\_ (if applicable).

I hereby state that I have reviewed and understand the content of the above-identified specification, including the claims, as amended by any amendment(s) referred to above.

I acknowledge my duty to disclose to the Patent and Trademark Office all information known to me to be material to patentability as defined in Title 37, Code of Federal Regulations, ' 1.56.

I hereby claim priority benefits under Title 35, United States Code, ' 119 (and ' 172 if this application is for a Design) of any application(s) for patent or inventor's certificate listed below and have also identified below any application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

COUNTRY	APPLICATION NO.	DATE OF FILING	PRIORITY CLAIMED
Norway	19994381	September 10, 1999	Yes


I hereby claim the benefit under Title 35, United States Code ' 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code ' 112, I acknowledge the duty to disclose information material to patentability as defined in Title 37, Code of Federal Regulations, ' 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application

APPLICATION SERIAL NO.	U.S. FILING DATE	STATUS: PATENTED, PENDING, ABANDONED

And I hereby appoint Michael R. Davis, Reg. No. 25,134; Matthew M. Jacob, Reg. No. 25,154; Warren M. Check, Jr., Reg. No. 33,367; Nils Pedersen, Reg. No. 33,145; Charles R. Watts, Reg. No. 33,142; and Michael S. Huppert, Reg. No. 40,268, who together constitute the firm

of WENDEROTH, LIND & PONACK, L.L.P., as well as any other attorneys and agents associated with Customer No. 000513, to prosecute this application and to transact all business in the U.S. Patent and Trademark Office connected therewith.

I hereby authorize the U.S. attorneys and agents named herein to accept and follow instructions from \_\_\_\_\_, as to any action to be taken in the U.S. Patent and Trademark Office regarding this application without direct communication between the U.S. attorneys and myself. In the event of a change in the persons from whom instructions may be taken, the U.S. attorneys named herein will be so notified by me.

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<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

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<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

<b>Full Name of Fourth Inventor</b>	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
<b>Residence &amp; Citizenship</b>	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

<b>Full Name of Fifth Inventor</b>	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
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<b>Residence &amp; Citizenship</b>	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

<b>Full Name of Sixth Inventor</b>	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
<b>Residence &amp; Citizenship</b>	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

<b>Full Name of Seventh Inventor</b>	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
<b>Residence &amp; Citizenship</b>	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
<b>Post Office Address</b>	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

I further declare that all statements made herein of my own knowledge are true, and that all statements on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Egil  
 1st Inventor Lundberg Egil Lundberg Date 7.3.02  
 2nd Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 3rd Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 4th Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 5th Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 6th Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 7th Inventor \_\_\_\_\_ Date \_\_\_\_\_

The above application may be more particularly identified as follows:

U.S. Application Serial No 10/070,374 Filing Date March 6, 2002  
 Applicant Reference Number P99070 AB: AMH Atty Docket No. 2002-0266A  
 Title of Invention \_\_\_\_\_